

3-[(Z)-(4-Diethylamino-6-oxocyclohexa-2,4-dien-1-ylidene)methylamino]benzoic acid

M. T. Swamy,^a B. Narayana,^b H. S. Yathirajan,^c B. K. Sarojini^d and Maciej Kubicki^{e*}

^aDepartment of Chemistry, Sambhram Institute of Technology, Bangalore 560 098, India, ^bDepartment of Studies in Chemistry, Mangalore University, Mangalagangothri 574 199, India, ^cDepartment of Studies in Chemistry, University of Mysore, Manasagangothri, Mysore 570 006, India, ^dDepartment of Chemistry, P. A. College of Engineering, Nadupadavu, Mangalore 574 153, India, and ^eDepartment of Chemistry, Adam Mickiewicz University, Grunwaldzka 6, 60-780 Poznań, Poland
Correspondence e-mail: mkubicki@amu.edu.pl

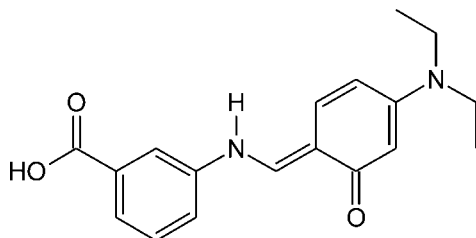
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Key indicators: single-crystal X-ray study; $T = 295$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.041; wR factor = 0.124; data-to-parameter ratio = 12.4.

The title compound, $\text{C}_{18}\text{H}_{20}\text{N}_2\text{O}_3$, crystallizes as the keto tautomer, unlike the vast majority of similar structures that have been reported that contain the hydroxy tautomer. There are two strong hydrogen bonds in the crystal structure, both accepted by the same carbonyl group: one intramolecular $\text{N}-\text{H}\cdots\text{O}$ and one intermolecular $\text{O}-\text{H}\cdots\text{O}$. As a result, the carbonyl $\text{C}=\text{O}$ distance is long, at 1.310 (2) Å, which may suggest the molecule has a significant zwitterionic character. The dihedral angle between the benzene ring planes is 15.05 (7)°. As a result of the intramolecular hydrogen bond, the bridging $\text{C}-\text{C}=\text{N}-\text{C}$ group is almost coplanar with the benzene ring that has the diethylamino substituent [dihedral angle 2.35 (15)°].

Related literature

For related structures, see: Büyükgüngör *et al.* (2007); Odabaşoğlu *et al.* (2007); Yathirajan *et al.* (2007). For biological applications, see Hodnett & Dunn (1970); Misra *et al.* (1981); Agarwal *et al.* (1983); Varma *et al.* (1986); Singh & Dash (1988). For related literature, see: Allen (2002).



Experimental

Crystal data

$\text{C}_{18}\text{H}_{20}\text{N}_2\text{O}_3$
 $M_r = 312.36$
Monoclinic, $P2_1/n$
 $a = 9.0904$ (13) Å
 $b = 9.8993$ (9) Å
 $c = 17.8208$ (19) Å
 $\beta = 100.068$ (2)°
 $V = 1579.0$ (3) Å³
 $Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 295$ (1) K
 $0.3 \times 0.2 \times 0.2$ mm

Data collection

KUMA KM4CCD diffractometer
Absorption correction: none
12993 measured reflections
3514 independent reflections
2464 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.020$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.041$
 $wR(F^2) = 0.124$
 $S = 0.98$
3514 reflections
283 parameters
H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.25$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.17$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N7}-\text{H7}\cdots\text{O10}$	0.91 (2)	1.81 (2)	2.578 (2)	140 (2)
$\text{O1A2}-\text{H1A2}\cdots\text{O10}^i$	1.07 (3)	1.40 (3)	2.467 (1)	173 (2)
$\text{C5}-\text{H5}\cdots\text{O1A1}^{ii}$	0.98 (2)	2.48 (2)	3.449 (2)	174 (2)
$\text{C8}-\text{H8}\cdots\text{O1A1}^{iii}$	0.96 (2)	2.63 (2)	3.436 (2)	142 (1)
$\text{C11}-\text{H11}\cdots\text{O1A2}^{iv}$	0.98 (2)	2.60 (2)	3.244 (2)	124 (1)
$\text{C16}-\text{H16A}\cdots\text{O1A1}^v$	1.00 (2)	2.60 (2)	3.570 (2)	163 (1)

Symmetry codes: (i) $x, y-1, z$; (ii) $-x+\frac{5}{2}, y+\frac{1}{2}, -z+\frac{5}{2}$; (iii) $-x+2, -y, -z+2$; (iv) $x, y+1, z$; (v) $x-\frac{1}{2}, -y+\frac{1}{2}, z-\frac{1}{2}$.

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2006); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2006); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* (Siemens, 1989); software used to prepare material for publication: *SHELXL97* and *WinGX* (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CS2063).

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supporting information

Acta Cryst. (2008). E64, o163–o164 [https://doi.org/10.1107/S1600536807064446]

3-[(Z)-(4-Diethylamino-6-oxocyclohexa-2,4-dien-1-ylidene)methylamino]-benzoic acid

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S1. Comment

Schiff bases are used as substrates in the preparation of number of industrial and biologically active compounds *via* ring closure, cycloaddition and replacement reactions. Some Schiff base derivatives were reported to possess antimicrobial, anti-inflammatory and central nervous system activities. Moreover, Schiff bases are also known to have biological activities such as antimicrobial, antifungal, antitumor and as herbicides (*e.g.* Hodnett *et al.*, 1970, Singh & Dash, 1988, Varma *et al.*, (1986)). In the course of our studies of Schiff bases (*e.g.* Büyükgüngör *et al.*, 2007; Odabaşoğlu *et al.*, 2007; Yathirajan *et al.*, 2007), the title compound, C₁₈H₂₀N₂O₃ was synthesized and its crystal structure is reported.

The title compound crystallizes as the keto-tautomer (Fig. 1), unlike the majority of the similar compounds. In the CSD (Allen, 2002) the hydroxy tautomers were found in 218 compounds while the keto ones only in 17 compounds (Version Nov. 2006, updates up to August 2007; only organic compounds). In this case the presence of the certain tautomer is proven by the successful location and refinement of the hydrogen atom bonded to N7 nitrogen atom (Fig. 2a). Short and relatively linear intramolecular N—H \cdots O hydrogen bond forms an almost planar (maximum deviation 0.017 (1) Å) six-membered ring. The same O10 oxygen atom which accepts the intramolecular hydrogen bond is involved in a very short (O \cdots O distance is 2.467 (1) Å) intermolecular O—H \cdots O hydrogen bond. As a result the C10—O10 bond of 1.310 (2) Å is significantly longer than a typical C=O double bond. CSD search results show that such elongation is typical for similar compounds, the mean C—O distance being 1.348 (17) Å for hydroxy tautomers, but it may also be as large as 1.299 (17) Å also for the keto-tautomers. Together with the observation of bond lengths around N7 atom this implies some degree of zwitterionic character of the molecule, with partial positive charge at N7—H7 group and negative at O10 atom. These perturbations disturb also the benzene ring. The ring A (C1 - C6) is closer to planarity than the ring B (C9 - C14). Maximum deviations from the least-squares planes are 0.0103 (11) Å for ring A and 0.0219 (12) Å for ring B. Also the bond lengths and angles are much more uniform within the ring A than in ring B.

The conformation of the molecule is described by the dihedral angles between benzene ring planes of 15.05 (7)°. As a result of the intramolecular hydrogen bond, the bridging C—C=N—C group is almost coplanar with the ring B (dihedral angle 2.35 (15)°). The COO group is significantly, by 21.73 (10)°, twisted with respect to its parent ring's plane. On the other end of the molecule, the C—N—C fragment is twisted by 17.0 (2)°, while the terminal C—C bonds are almost perpendicular to the CNC plane.

In the crystal structure the molecules are connected by strong intermolecular O—H \cdots O hydrogen bonds into tapes along the *y*-direction. The O—H bond is significantly elongated, to 1.07 (3) Å due to the formation of the hydrogen bond (*cf.* Fig. 2 b). The tapes are connected by relatively strong inter-tape C—H \cdots O hydrogen bonds into the layers (Fig. 3). Some additional C—H \cdots O interactions (Table 1) also play a role in the building of the crystal structure.

S2. Experimental

A mixture of 3-aminobenzoic acid (1.37 g, 0.01 mol) and 4-(diethylamino)-2-hydroxybenzaldehyde (1.92 g, 0.01 mol) in 25 ml of absolute ethanol containing 2 drops of 4 M sulfuric acid was refluxed for about 5 h. On cooling, the separated solid was filtered and recrystallized from DMF (m.p.: 483–485 K). The expected product was 3-((1E)-[4-(diethylamino)-2-hydroxyphenyl]methylene)amino)benzoic acid, but the obtained product was the tautomeric form 3-((Z)-[4-(diethylamino)-6-oxocyclohexa-2,4-dien-1-ylidene]methyl) amino)benzoic acid. Analysis for $C_{18}H_{20}N_2O_3$: Found (Calculated): C: 69.12 (69.21); H: 6.38 (6.45); N: 8.90% (8.97%).

S3. Refinement

The hydrogen atoms were located in the difference Fourier maps and refined as 'riding model'. Isotropic displacement parameters for hydrogen atoms were set at 1.2 (1.3 for methyl group) times the U_{eq} values of appropriate carrier atoms.

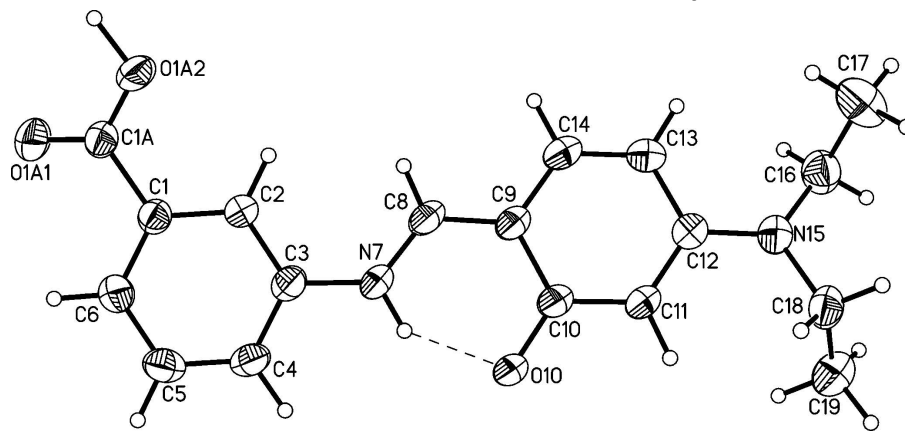


Figure 1

Anisotropic displacement ellipsoids representation (50% probability level) of the molecule with the atom labelling scheme, intramolecular hydrogen bond is depicted in dashed line.

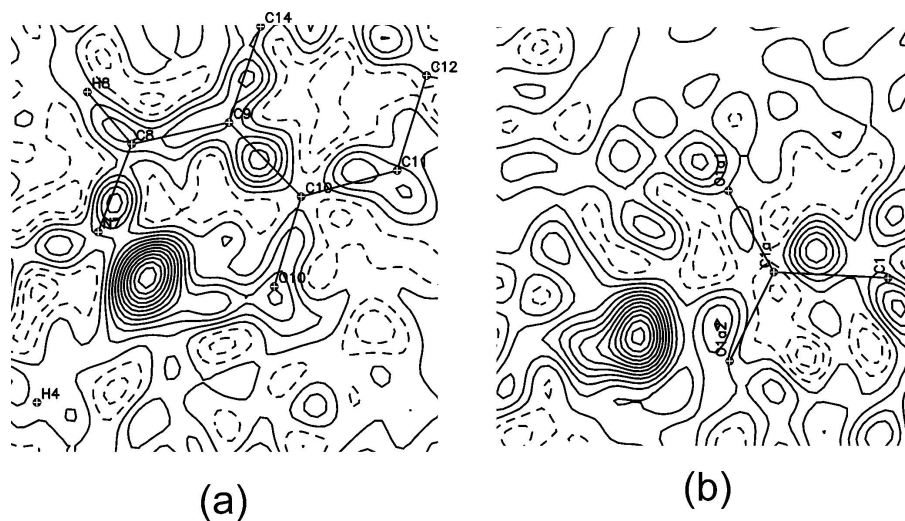


Figure 2

Difference Fourier map slices calculated for a model without the hydrogen atoms involved in intramolecular hydrogen bonds: (a) H7, (b) H1A2 (Farrugia, 1999). Solid lines: positive values, dashed: negative; contour level: 0.04 $e\%A^{-3}$.

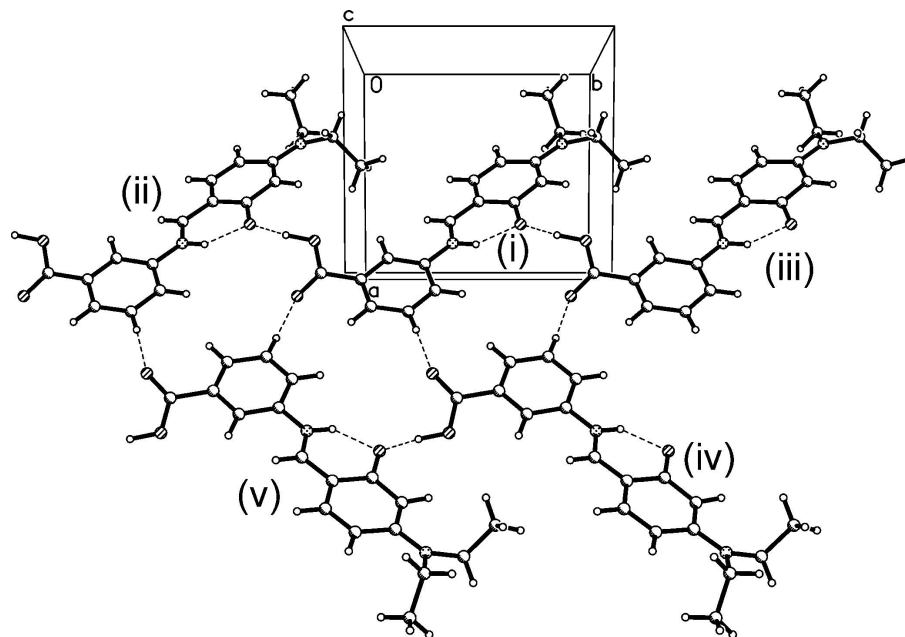


Figure 3

The hydrogen-bonded layer as seen approximately along the *c* axis. Symmetry codes: (i) x,y,z (ii) $x,-1+y,z$ (iii) $x,1+y,z$ (iv) $5/2-x,1/2+y,5/2-z$ (v) $5/2-x,-1/2+y,5/2-z$.

3-[(Z)-(4-Diethylamino-6-oxocyclohexa-2,4-dien-1-ylidene)methylamino]benzoic acid

Crystal data

$C_{18}H_{20}N_2O_3$

$M_r = 312.36$

Monoclinic, $P2_1/n$

Hall symbol: $-P\ 2_1n$

$a = 9.0904\ (13)\ \text{\AA}$

$b = 9.8993\ (9)\ \text{\AA}$

$c = 17.8208\ (19)\ \text{\AA}$

$\beta = 100.068\ (2)^\circ$

$V = 1579.0\ (3)\ \text{\AA}^3$

$Z = 4$

$F(000) = 664$

$D_x = 1.314\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 5056 reflections

$\theta = 3\text{--}24^\circ$

$\mu = 0.09\ \text{mm}^{-1}$

$T = 295\ \text{K}$

Block, purple

$0.3 \times 0.2 \times 0.2\ \text{mm}$

Data collection

KUMA KM4CCD four-circle

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: $8.1929\ \text{pixels mm}^{-1}$

ω scan

12993 measured reflections

3514 independent reflections

2464 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.020$

$\theta_{\text{max}} = 28.0^\circ$, $\theta_{\text{min}} = 2.4^\circ$

$h = -11 \rightarrow 11$

$k = -12 \rightarrow 12$

$l = -23 \rightarrow 21$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.041$

$wR(F^2) = 0.124$

$S = 0.98$

3514 reflections

283 parameters

0 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from
neighbouring sites

H atoms treated by a mixture of independent
and constrained refinement

$$w = 1/[\sigma^2(F_o^2) + (0.063P)^2 + 0.4649P]$$

$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\max} = 0.003$$

$$\Delta\rho_{\max} = 0.25 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.17 \text{ e } \text{\AA}^{-3}$$

Extinction correction: *SHELXL*,

$$F_c^* = kFc[1 + 0.001x Fc^2 \lambda^3 / \sin(2\theta)]^{-1/4}$$

Extinction coefficient: 0.0026 (8)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso} */ <i>U</i> _{eq}
C1	1.03151 (16)	0.06899 (14)	1.12525 (8)	0.0326 (3)
C1A	1.00497 (17)	−0.07858 (14)	1.10991 (8)	0.0353 (3)
O1A1	1.09883 (14)	−0.16365 (11)	1.13287 (7)	0.0535 (3)
O1A2	0.87299 (13)	−0.10532 (11)	1.07198 (7)	0.0509 (3)
H1A2	0.850 (3)	−0.211 (3)	1.0667 (14)	0.098 (8)*
C2	0.94453 (17)	0.16506 (14)	1.08096 (8)	0.0330 (3)
H2	0.8663 (17)	0.1354 (16)	1.0377 (9)	0.035 (4)*
C3	0.96859 (17)	0.30147 (14)	1.09759 (8)	0.0347 (3)
C4	1.07595 (18)	0.34076 (15)	1.15860 (9)	0.0401 (4)
H4	1.0907 (18)	0.4382 (18)	1.1696 (9)	0.043 (4)*
C5	1.16149 (19)	0.24488 (17)	1.20297 (9)	0.0432 (4)
H5	1.234 (2)	0.2745 (18)	1.2469 (10)	0.050 (5)*
C6	1.14098 (17)	0.10890 (16)	1.18570 (9)	0.0388 (3)
H6	1.2029 (19)	0.0408 (19)	1.2189 (10)	0.048 (5)*
N7	0.88185 (16)	0.40373 (12)	1.05574 (7)	0.0401 (3)
H7	0.887 (2)	0.489 (2)	1.0747 (11)	0.065 (6)*
C8	0.78879 (19)	0.39282 (15)	0.99049 (9)	0.0411 (4)
H8	0.7807 (19)	0.3078 (19)	0.9646 (10)	0.051 (5)*
C9	0.70296 (19)	0.49992 (14)	0.95604 (9)	0.0398 (4)
C10	0.71045 (18)	0.63293 (14)	0.99024 (8)	0.0372 (3)
O10	0.80433 (15)	0.65435 (10)	1.05372 (6)	0.0485 (3)
C11	0.62075 (19)	0.73538 (15)	0.95329 (9)	0.0392 (4)
H11	0.6353 (18)	0.8253 (17)	0.9767 (9)	0.041 (4)*
C12	0.53163 (17)	0.71697 (15)	0.88196 (9)	0.0388 (4)
C13	0.5272 (2)	0.58514 (17)	0.84767 (10)	0.0482 (4)
H13	0.466 (2)	0.5678 (18)	0.7969 (11)	0.054 (5)*
C14	0.6085 (2)	0.48282 (16)	0.88451 (10)	0.0482 (4)
H14	0.602 (2)	0.392 (2)	0.8607 (10)	0.059 (5)*
N15	0.45111 (16)	0.82021 (13)	0.84509 (8)	0.0449 (3)
C16	0.3912 (2)	0.8173 (2)	0.76261 (10)	0.0495 (4)
H16A	0.457 (2)	0.7599 (18)	0.7356 (10)	0.054 (4)*
H16B	0.394 (2)	0.908 (2)	0.7398 (10)	0.054 (4)*
C17	0.2318 (2)	0.7704 (3)	0.74366 (13)	0.0670 (6)
H17A	0.228 (3)	0.675 (3)	0.7658 (15)	0.098 (5)*
H17B	0.200 (3)	0.773 (3)	0.6867 (16)	0.098 (5)*
H17C	0.161 (3)	0.833 (3)	0.7715 (15)	0.098 (5)*
C18	0.4280 (2)	0.94628 (17)	0.88460 (10)	0.0442 (4)
H18A	0.4192 (18)	0.9237 (17)	0.9378 (10)	0.045 (3)*

H18B	0.334 (2)	0.9777 (17)	0.8604 (10)	0.045 (3)*
C19	0.5467 (2)	1.05246 (19)	0.88190 (13)	0.0552 (5)
H19A	0.647 (2)	1.021 (2)	0.9066 (12)	0.071 (4)*
H19B	0.523 (2)	1.138 (2)	0.9093 (12)	0.071 (4)*
H19C	0.549 (2)	1.075 (2)	0.8283 (13)	0.071 (4)*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0348 (8)	0.0296 (7)	0.0333 (7)	0.0018 (6)	0.0059 (6)	0.0005 (6)
C1A	0.0403 (8)	0.0294 (7)	0.0354 (7)	0.0035 (6)	0.0047 (6)	0.0014 (6)
O1A1	0.0530 (7)	0.0334 (6)	0.0680 (8)	0.0117 (5)	−0.0068 (6)	0.0036 (5)
O1A2	0.0482 (7)	0.0247 (6)	0.0722 (8)	0.0033 (5)	−0.0105 (6)	−0.0060 (5)
C2	0.0366 (8)	0.0272 (7)	0.0338 (7)	−0.0008 (6)	0.0021 (6)	−0.0013 (5)
C3	0.0417 (8)	0.0271 (7)	0.0361 (7)	−0.0003 (6)	0.0090 (6)	0.0001 (6)
C4	0.0453 (9)	0.0298 (8)	0.0445 (9)	−0.0059 (6)	0.0061 (7)	−0.0067 (6)
C5	0.0408 (9)	0.0453 (9)	0.0410 (9)	−0.0053 (7)	0.0001 (7)	−0.0081 (7)
C6	0.0365 (8)	0.0386 (8)	0.0395 (8)	0.0031 (7)	0.0017 (6)	0.0005 (6)
N7	0.0538 (8)	0.0219 (6)	0.0418 (7)	0.0012 (6)	0.0009 (6)	0.0000 (5)
C8	0.0562 (10)	0.0224 (7)	0.0436 (8)	−0.0012 (7)	0.0053 (7)	−0.0022 (6)
C9	0.0525 (9)	0.0240 (7)	0.0405 (8)	−0.0010 (7)	0.0020 (7)	−0.0010 (6)
C10	0.0470 (9)	0.0256 (7)	0.0371 (8)	−0.0022 (6)	0.0022 (7)	−0.0021 (6)
O10	0.0707 (8)	0.0234 (5)	0.0433 (6)	0.0011 (5)	−0.0131 (6)	−0.0034 (4)
C11	0.0502 (9)	0.0243 (7)	0.0407 (8)	0.0022 (6)	0.0009 (7)	−0.0047 (6)
C12	0.0407 (9)	0.0317 (7)	0.0423 (8)	0.0012 (6)	0.0021 (7)	−0.0018 (6)
C13	0.0555 (11)	0.0379 (9)	0.0449 (9)	−0.0005 (8)	−0.0087 (8)	−0.0082 (7)
C14	0.0636 (11)	0.0276 (8)	0.0489 (9)	−0.0003 (7)	−0.0026 (8)	−0.0076 (7)
N15	0.0495 (8)	0.0375 (7)	0.0436 (7)	0.0080 (6)	−0.0031 (6)	−0.0014 (6)
C16	0.0528 (11)	0.0497 (10)	0.0452 (9)	0.0063 (8)	0.0059 (8)	0.0031 (8)
C17	0.0530 (12)	0.0885 (17)	0.0554 (12)	0.0000 (11)	−0.0018 (10)	−0.0032 (11)
C18	0.0447 (10)	0.0390 (8)	0.0480 (9)	0.0116 (7)	0.0058 (8)	0.0012 (7)
C19	0.0618 (12)	0.0410 (10)	0.0636 (12)	0.0016 (9)	0.0128 (10)	−0.0003 (9)

Geometric parameters (\AA , $^\circ$)

C1—C6	1.390 (2)	C10—C11	1.393 (2)
C1—C2	1.391 (2)	C11—C12	1.395 (2)
C1—C1A	1.498 (2)	C11—H11	0.982 (17)
C1A—O1A1	1.2177 (18)	C12—N15	1.3578 (19)
C1A—O1A2	1.2972 (19)	C12—C13	1.439 (2)
O1A2—H1A2	1.07 (3)	C13—C14	1.354 (2)
C2—C3	1.392 (2)	C13—H13	0.991 (18)
C2—H2	0.998 (16)	C14—H14	1.00 (2)
C3—C4	1.384 (2)	N15—C18	1.466 (2)
C3—N7	1.4132 (19)	N15—C16	1.476 (2)
C4—C5	1.384 (2)	C16—C17	1.503 (3)
C4—H4	0.989 (17)	C16—H16A	1.004 (19)
C5—C6	1.386 (2)	C16—H16B	0.990 (19)

C5—H5	0.975 (18)	C17—H17A	1.03 (3)
C6—H6	1.002 (18)	C17—H17B	1.01 (3)
N7—C8	1.317 (2)	C17—H17C	1.07 (3)
N7—H7	0.91 (2)	C18—C19	1.513 (3)
C8—C9	1.394 (2)	C18—H18A	0.990 (18)
C8—H8	0.956 (19)	C18—H18B	0.942 (18)
C9—C14	1.417 (2)	C19—H19A	0.99 (2)
C9—C10	1.448 (2)	C19—H19B	1.02 (2)
C10—O10	1.3097 (18)	C19—H19C	0.99 (2)
C6—C1—C2	120.30 (13)	N15—C12—C11	121.38 (14)
C6—C1—C1A	119.30 (13)	N15—C12—C13	120.31 (14)
C2—C1—C1A	120.38 (13)	C11—C12—C13	118.31 (14)
O1A1—C1A—O1A2	124.32 (14)	C14—C13—C12	120.03 (15)
O1A1—C1A—C1	122.15 (14)	C14—C13—H13	119.0 (11)
O1A2—C1A—C1	113.51 (13)	C12—C13—H13	120.9 (11)
C1A—O1A2—H1A2	113.4 (13)	C13—C14—C9	122.70 (15)
C1—C2—C3	119.34 (14)	C13—C14—H14	119.1 (11)
C1—C2—H2	119.7 (9)	C9—C14—H14	118.2 (11)
C3—C2—H2	121.0 (9)	C12—N15—C18	121.24 (13)
C4—C3—C2	120.21 (14)	C12—N15—C16	122.74 (14)
C4—C3—N7	117.72 (13)	C18—N15—C16	115.87 (13)
C2—C3—N7	122.02 (14)	N15—C16—C17	113.77 (16)
C3—C4—C5	120.31 (14)	N15—C16—H16A	110.0 (10)
C3—C4—H4	118.8 (9)	C17—C16—H16A	110.3 (10)
C5—C4—H4	120.8 (10)	N15—C16—H16B	111.0 (11)
C4—C5—C6	119.92 (15)	C17—C16—H16B	106.3 (11)
C4—C5—H5	119.0 (11)	H16A—C16—H16B	105.0 (15)
C6—C5—H5	121.0 (11)	C16—C17—H17A	106.9 (15)
C5—C6—C1	119.89 (15)	C16—C17—H17B	108.2 (15)
C5—C6—H6	118.8 (10)	H17A—C17—H17B	113 (2)
C1—C6—H6	121.2 (10)	C16—C17—H17C	110.5 (14)
C8—N7—C3	128.32 (13)	H17A—C17—H17C	107 (2)
C8—N7—H7	113.0 (13)	H17B—C17—H17C	112 (2)
C3—N7—H7	118.7 (13)	N15—C18—C19	114.44 (15)
N7—C8—C9	123.38 (14)	N15—C18—H18A	108.1 (10)
N7—C8—H8	118.8 (11)	C19—C18—H18A	111.4 (10)
C9—C8—H8	117.8 (11)	N15—C18—H18B	104.8 (11)
C8—C9—C14	120.66 (14)	C19—C18—H18B	110.6 (11)
C8—C9—C10	121.66 (14)	H18A—C18—H18B	107.1 (14)
C14—C9—C10	117.65 (14)	C18—C19—H19A	111.7 (13)
O10—C10—C11	121.99 (13)	C18—C19—H19B	110.3 (12)
O10—C10—C9	119.11 (13)	H19A—C19—H19B	107.8 (18)
C11—C10—C9	118.86 (14)	C18—C19—H19C	109.2 (13)
C10—C11—C12	122.32 (13)	H19A—C19—H19C	109.2 (17)
C10—C11—H11	115.8 (10)	H19B—C19—H19C	108.5 (17)
C12—C11—H11	121.3 (9)		

C6—C1—C1A—O1A1	−21.1 (2)	C14—C9—C10—O10	−175.35 (15)
C2—C1—C1A—O1A1	160.84 (14)	C8—C9—C10—C11	−179.40 (15)
C6—C1—C1A—O1A2	157.39 (14)	C14—C9—C10—C11	2.7 (2)
C2—C1—C1A—O1A2	−20.7 (2)	O10—C10—C11—C12	173.43 (15)
C6—C1—C2—C3	0.2 (2)	C9—C10—C11—C12	−4.5 (2)
C1A—C1—C2—C3	178.28 (13)	C10—C11—C12—N15	−176.43 (16)
C1—C2—C3—C4	−1.3 (2)	C10—C11—C12—C13	3.3 (3)
C1—C2—C3—N7	−178.63 (13)	N15—C12—C13—C14	179.57 (17)
C2—C3—C4—C5	0.8 (2)	C11—C12—C13—C14	−0.1 (3)
N7—C3—C4—C5	178.23 (14)	C12—C13—C14—C9	−1.6 (3)
C3—C4—C5—C6	0.8 (2)	C8—C9—C14—C13	−177.61 (18)
C4—C5—C6—C1	−1.9 (2)	C10—C9—C14—C13	0.3 (3)
C2—C1—C6—C5	1.4 (2)	C11—C12—N15—C18	−13.4 (2)
C1A—C1—C6—C5	−176.68 (14)	C13—C12—N15—C18	166.90 (16)
C4—C3—N7—C8	170.02 (16)	C11—C12—N15—C16	162.02 (16)
C2—C3—N7—C8	−12.6 (3)	C13—C12—N15—C16	−17.7 (3)
C3—N7—C8—C9	176.68 (15)	C12—N15—C16—C17	95.6 (2)
N7—C8—C9—C14	177.89 (16)	C18—N15—C16—C17	−88.7 (2)
N7—C8—C9—C10	0.0 (3)	C12—N15—C18—C19	89.8 (2)
C8—C9—C10—O10	2.6 (2)	C16—N15—C18—C19	−86.0 (2)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N7—H7 \cdots O10	0.91 (2)	1.81 (2)	2.578 (2)	140 (2)
O1A2—H1A2 \cdots O10 ⁱ	1.07 (3)	1.40 (3)	2.467 (1)	173 (2)
C5—H5 \cdots O1A1 ⁱⁱ	0.98 (2)	2.48 (2)	3.449 (2)	174 (2)
C8—H8 \cdots O1A1 ⁱⁱⁱ	0.96 (2)	2.63 (2)	3.436 (2)	142 (1)
C11—H11 \cdots O1A2 ^{iv}	0.98 (2)	2.60 (2)	3.244 (2)	124 (1)
C16—H16A \cdots O1A1 ^v	1.00 (2)	2.60 (2)	3.570 (2)	163 (1)

Symmetry codes: (i) $x, y-1, z$; (ii) $-x+5/2, y+1/2, -z+5/2$; (iii) $-x+2, -y, -z+2$; (iv) $x, y+1, z$; (v) $x-1/2, -y+1/2, z-1/2$.